

SHORT COMMUNICATION

TRITERPENES FROM *THUIDIUM TAMARISCIFOLIUM**

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Abstract—Nine triterpenes and sterols have been isolated from *Thuidium tamariscifolium*. A number of aliphatic hydrocarbons and fatty acids were also detected.

FROM the light petroleum extract of the moss *Thuidium tamariscifolium* (Neck.) Lindb. (= *Thuidium tamariscinum* B. e.) the following triterpenes and sterols were isolated and identified: 22(29)-Hopene,² 7-fernene,^{2c, 3} 9(11)-fernene,^{2b, 3, 4} 24-methylenecycloartanol,⁵ lupeol, ursolic acid, campesterol (identified by GLC), stigmasterol (identified by GLC) and β -sitosterol. Moreover, aliphatic hydrocarbons from C₂₂ to C₃₃ were also found in the plant and, after saponification of the extract, fatty acids from C₁₄ to C₂₂.

In our first communication on this subject,¹ we remarked that hopane hydrocarbons seem to be characteristic of primitive plants; their occurrence in ferns and mosses also points to a possible phylogenetic connexion between bryophytes and pteridophytes.^{6, 7} Further studies, now in progress, seem to support this hypothesis.

EXPERIMENTAL

M.ps were determined with a Kofler apparatus and are uncorrected. Infrared spectra were registered on nujol mulls with a Perkin-Elmer Infracord, Mod. 137 spectrophotometer. GLC analyses were performed with the following instruments: Perkin-Elmer F 20, for triterpenes and sterol trimethylsilyl ethers (column, 3 per cent SE 52 silicone rubber on Chromosorb W 80–100 mesh, temp. 240°, injection block temp. 260°, carrier gas N₂, flow rate 50 ml/min); Carlo Erba Fractovap, Mod. G.V., for aliphatic hydrocarbons (column, 1 per cent neopentyl glycol succinate on Chromosorb W 80–100 mesh, programmed temp. from 190° to 230°, increase 1.5°/min, injection block temp. 240°, carrier gas N₂, flow rate from 39 to 32 ml/min); Perkin-Elmer 800, for fatty acid methyl esters (column, 20 per cent polyethylene glycol succinate on acid-washed Chromosorb W 80–100 mesh, temp. 200°, injection block temp. 290°, carrier gas N₂, flow rate 40 ml/min). Specific rotations were measured with a Perkin-Elmer, Mod. 141 photoelectric polarimeter, at 25°. Light petroleum refers to the fraction of boiling range 30–60°. Identities of products were determined by comparison of i.r. spectra and GLC retention times.

* "Triterpenes from Mosses-II". For Part I see Ref. 1.

¹ Part I, A. MARSILI and I. MORELLI, *Phytochem.* 7, 1705 (1968).

² (a) W. J. DUNSTAN, H. FAZAKERLEY, T. G. HALSALL and E. R. H. JONES, *Croat. Chem. Acta* 29, 173 (1957);

(b) H. AGETA, K. IWATA and K. YONEZAWA, *Chem. Pharm. Bull. Japan* 11, 408 (1963); (c) H. AGETA, K. IWATA and S. NATORI, *Tetrahedron Letters* 3413 (1964).

³ H. AGETA and K. IWATA, *Tetrahedron Letters* 6069 (1966).

⁴ H. AGETA, K. IWATA and S. NATORI, *Tetrahedron Letters* 1447 (1963).

⁵ Triterpenoids of the cyclolanostane group were also found in ferns. See, e.g., G. BERTI, F. BOTTARI, B. MACCHIA, A. MARSILI, G. OURISSON and H. PIOTROWSKA, *Bull. soc. chim. Fr.* 2359 (1964).

⁶ See G. M. SMITH, *Cryptogamic Botany*, Vol. II, pp. 131–134, McGraw-Hill, New York (1955).

⁷ G. HASKALL [*Bryologist* 52, 49 (1949)] holds that bryophytes descend from pteridophytes.

Extraction

The dried plant (3 kg), collected on April 1968 in a wood of *Castanea sativa* Mill., was extracted in a Soxhlet with light petroleum (8 l.) for 45 hr. On cooling a ppt (0.25 g) separated out; this was combined with the ppt (0.27 g) formed when the filtrate was evaporated to 2 l. Crystallization of the solid from EtOH afforded ursolic acid.⁸ Further concentration of the extract to 250 ml gave waxes (3.1 g), m.p. 75–81°, after crystallization from benzene–MeOH; λ_{CO} 5.80 μ . The remaining soln. was evaporated to dryness to give a brown residue (31 g) which was refluxed with 10 per cent NaOH–EtOH (200 ml) for 4 hr. After this time the solvent was removed by distillation and the residue, dried at 90°, was extracted in a Soxhlet with light petroleum (300 ml) for 100 hr. The extract, on evaporation, afforded 6 g of unsaponifiable material. The unextracted solid was taken up in water and the soln. was acidified with conc. HCl and extracted with Et₂O. Evaporation of the Et₂O afforded 15 g of fatty acids, 1 g of which was converted into the corresponding methyl esters with CH₂N₂.

Unsaponifiable Material

A part of the unsaponifiable fraction (5 g) was dissolved in light petroleum and chromatographed over neutral alumina (Fluka, grade I, 200 g, 2 × 65 cm column). Light petroleum eluted hydrocarbons (aliphatic and triterpenoidic); no residue was obtained with benzene; Et₂O eluted mixtures of alcohols. The combined hydrocarbon fractions (0.4 g) were re-chromatographed over silica gel (Woelm, grade I, 120 g, 2 × 52 cm column; 20-ml fractions) impregnated with 15 per cent of AgNO₃. Light petroleum eluted aliphatic hydrocarbons (0.12 g), 9(11)-farnene (5 mg) and 7-farnene (35 mg); benzene eluted 22(29)-hopene (0.12 g). The alcoholic fractions, analysed by GLC (as trimethylsilyl ethers) contained at least eight components (mainly campesterol, stigmasterol, β -sitosterol and 24-methylenecycloartanol). A fraction (1.1 g) showing in the i.r. spectrum a band at 11.25 μ (>C=CH_2) was benzoated; the resulting mixture of benzoates was chromatographed over silica gel–AgNO₃ (110 g, 2 × 50 cm column; 20-ml fractions), using as eluant light petroleum containing increasing amounts of benzene. The fractions obtained with light petroleum–benzene (1:1) were combined and the residue (0.12 g) crystallized from light petroleum. Lupeol benzoate⁸ (30 mg) separated out. The mother liquor, after evaporation, left a residue which was crystallized from acetone–MeOH; 24-methylenecycloartanol benzoate (20 mg) separated out. Another alcoholic fraction (0.5 g) containing campesterol (30 per cent), stigmasterol (23 per cent), β -sitosterol (38 per cent) and another unidentified component, was converted into the benzoates and chromatographed as the preceding fraction. Elution with light petroleum–benzene (9:1) afforded pure β -sitosterol benzoate (0.1 g).

Compounds

Ursolic acid, m.p. 284–288° (from EtOH), $[\alpha]_D + 65.3^\circ$ (c 1, EtOH). (Found: C, 78.75; H, 10.34. Calc. for C₃₀H₄₈O₃: C, 78.89; H, 10.59%). Lit.⁹ m.p. 289–290°, $[\alpha]_D + 64^\circ$.

9(11)-*Farnene*, m.p. 168–170°, $[\alpha]_D - 16^\circ$ (c 0.4, CHCl₃). Lit.⁴ m.p. 170–171°, $[\alpha]_D - 16.5^\circ$.

7-*Farnene*, m.p. 208–210° (from hexane), $[\alpha]_D - 26.8^\circ$ (c 1, CHCl₃). (Found: C, 87.60; H, 12.47. Calc. for C₃₀H₅₀: C, 87.73; H, 12.27%). Lit.^{2c} m.p. 208.5–209.5°, $[\alpha]_D - 27^\circ$.

22(29)-*Hopene*, m.p. 209–211° (from acetone), $[\alpha]_D + 60.3^\circ$ (c 1, CHCl₃). (Found: C, 87.55; H, 12.38. Calc. for C₃₀H₅₀: C, 87.73; H, 12.27%). Lit.^{2c} m.p. 210–211°, $[\alpha]_D + 61^\circ$.

Lupeol benzoate, m.p. 258–261° (from hexane), $[\alpha]_D + 62.5^\circ$ (c 0.8, CHCl₃). Found: C, 83.97; H, 10.40. Calc. for C₃₇H₅₄O₂: C, 83.72; H, 10.25%. Lit.¹⁰ m.p. 265°, $[\alpha]_D + 60.1^\circ$.

24-*Methylenecycloartanol benzoate*, m.p. 154–156° (from acetone–MeOH), $[\alpha]_D + 63.5^\circ$ (c 0.7, CHCl₃). (Found: C, 83.54; H, 10.30. Calc. for C₃₈H₅₆O₂: C, 83.77; H, 10.36%). Lit.¹¹ m.p. 156–157°, $[\alpha]_D + 62^\circ$.

β -*Sitosterol benzoate*, m.p. 146–147° (from hexane), $[\alpha]_D - 13.5^\circ$ (c 1, CHCl₃). (Found: C, 83.20; H, 10.24. Calc. for C₃₆H₅₄O₂: C, 83.34; H, 10.49%). Lit.¹² m.p. 146–147°, $[\alpha]_D - 13.8^\circ$.

β -*Sitosterol*, m.p. 140–142° (from chloroform–MeOH), $[\alpha]_D - 36.5^\circ$ (c 1, CHCl₃). (Found: C, 83.75; H, 12.30. Calc. for C₂₉H₅₀O: C, 83.99; H, 12.15%). Lit.¹² m.p. 136–137°, $[\alpha]_D - 36.6^\circ$.

Aliphatic hydrocarbons (GLC, per cent), C₂₃ (0.8); C₂₄ (0.5); C₂₅ (1.7); C₂₆ (0.9); C₂₇ (6.1); C₂₈ (1.6); C₂₉ (25.9); C₃₀ (2.9); C₃₁ (41.5); C₃₂ (3.5); C₃₃ (14.6).

⁸ Although it cannot be ruled out completely that ursolic acid and lupeol might have been derived from the *humus* in which the moss grew, since chestnut leaves contain these two triterpenes along with betulin (results from this laboratory); however, the plant material was well freed from the last traces of *Castanea* leaves, so that it seems highly probable that lupeol and ursolic acid are indeed present in the moss. Moreover, other mosses presently under study also appear to contain ursolic acid.

⁹ H. R. ARTHUR and W. H. HUI, *J. Chem. Soc.* 2782 (1954).

¹⁰ I. M. HEILBRON, T. KENNEDY and F. S. SPRING, *J. Chem. Soc.* 329 (1938).

¹¹ G. OHTA and M. SHIMIZU, *Chem. Pharm. Bull. Japan* 6, 325 (1958).

¹² E. S. WALLIS and P. N. CHAKRAVORTY, *J. Org. Chem.* 2, 335 (1937).

Fatty acids (GLC of methyl esters, per cent), myristic (1.0); pentadecanoic (1.0); palmitic (24.5); palmitoleic (2.0); heptadecanoic (1.0); heptadecenoic (0.5); stearic (3.0); oleic (13.0); linoleic (34.0); linolenic + arachidic (16.0); behenic (4.0).

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